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Key indicators

Single-crystal X-ray study
T = 293 K
Mean $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$
R factor = 0.055
wR factor = 0.178
Data-to-parameter ratio = 16.6

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

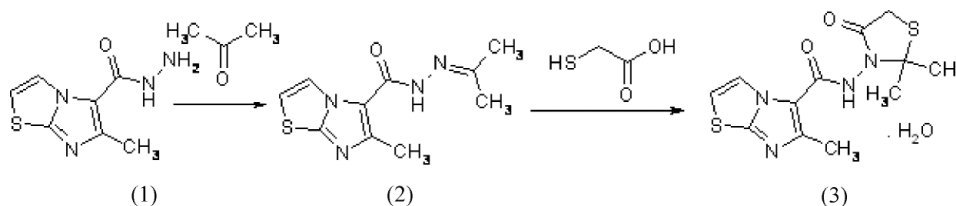
N-(2,2-Dimethyl-4-oxo-1,3-thiazolidin-3-yl)-6-methylimidazo[2,1-b][1,3]thiazole-5-carboxamide mono-hydrate

The title compound, $\text{C}_{12}\text{H}_{14}\text{N}_4\text{O}_2\text{S}_2 \cdot \text{H}_2\text{O}$, is a member of a new series of imidazo[2,1-b]thiazoles. The thiazolidine ring system adopts an envelope conformation. The packing is stabilized by intra- and intermolecular hydrogen-bond interactions.

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Comment

Imidazothiazole derivatives have demonstrated a broad range of biological activities, including immunoregulatory (Devlin & Hargrave, 1989), antihelminthic (Marin *et al.*, 1992), antimicrobial (Ulusoy *et al.*, 1997), cardiotoxic (Andreani *et al.*, 1998) and anticancer (Andreani *et al.*, 1992). These biological functions of imidazothiazole derivatives stimulated our research interest and we have synthesized the title imidazothiazole derivative N-(2,2-dimethyl-4-oxo-1,3-thiazolidin-3-yl)-6-methylimidazo[2,1-b][1,3]thiazole-5-carboxamide mono-hydrate, (3).



The main geometric parameters of (3) are listed in Table 1 and the molecular structure is illustrated in Fig. 1. The thiazole and imidazole rings are essentially coplanar and the least-squares plane containing all the non-H atoms has a maximum deviation of 0.005 (1) Å for atom N2. There is a dihedral angle of 0.38 (1)° between the thiazole and imidazole rings. The thiazolidine ring system adopts an envelope conformation,

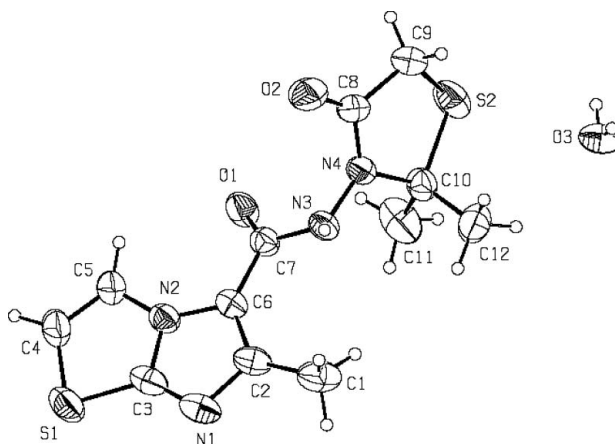


Figure 1 An ORTEP-3 drawing (Farrugia, 1997) of (3), with the atom-numbering scheme and 30% probability displacement ellipsoids.

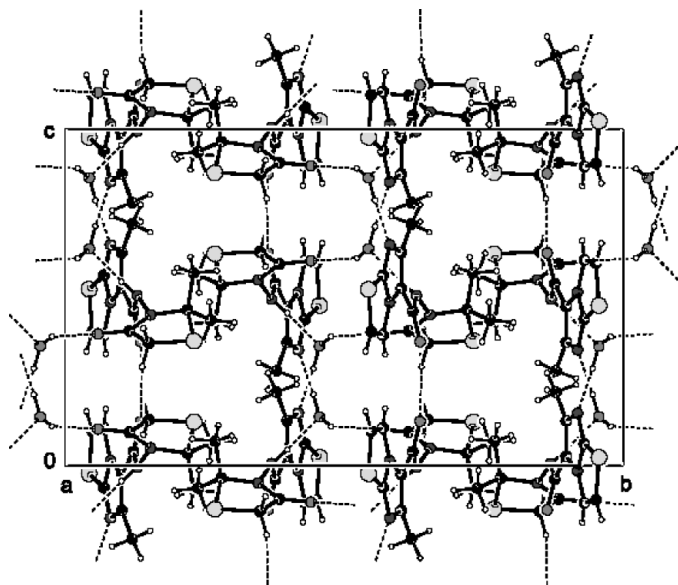


Figure 2
Packing and hydrogen-bond contacts (dashed lines) of (3).

with atom S2 0.173 (1) Å from the plane of the other four atoms. There is an angle of 72.46 (1)° between the imidazo[2,1-*b*][1,3]thiazole and thiazolidine ring systems.

The mean C—S bond length [1.7588 (5) Å] is longer than distances reported for similar molecules [1.729 (2) (Akkurt *et al.*, 2005) and 1.739 (5) Å (Vasu *et al.*, 2004)]. The other bond lengths and angles are in agreement with expected values (Allen *et al.*, 1987).

In the crystal structure, the molecules are linked by intra- and intermolecular hydrogen-bond contacts; relevant data are listed in Table 2 (Fig. 2).

Experimental

A mixture of 6-methylimidazo[2,1-*b*]thiazole-5-carboxylic acid isopropylidenehydrazide (1.18 g, 0.05 mol) and 2-mercaptoacetic acid (13.82 g, 0.15 mol) was refluxed in dry benzene (30 ml) for 6 h using a Dean–Stark trap. Excess benzene was evaporated *in vacuo*. The residue was triturated with saturated NaHCO₃ until CO₂ evolution ceased and allowed to stand overnight. The solid thus obtained was filtered off, washed with water and crystallized from a C₂H₅OH–H₂O mixture (Ur *et al.*, 2004) (m.p. 403–405 K). IR (KBr, cm⁻¹): 3312, 3125 (NH); 1691, 1663 (C=O). ¹H NMR (CDCl₃): 1.97 (3H, *s*, CH₃), 2.15 (3H, *s*, CH₃), 2.64 (3H, *s*, 6-CH₃), 6.89 (1H, *d*, *J* = 4.4 Hz, C₂-H), 8.24 (1H, *d*, *J* = 4.4 Hz, C₃-H), 8.34 (1H, *s*, CONH). EIMS (70 eV) *m/z* (%): 310 (*M*⁺, 43), 238 (1), 181 (6), 166 (30), 165 (100), 137 (8), 111 (6), 57 (12). Analysis calculated for C₁₂H₁₄N₄O₂S₂·H₂O: C 43.88, H 4.91, N 17.06%; found: C 44.15, H 5.20, N 17.00%.

Crystal data

C₁₂H₁₄N₄O₂S₂·H₂O
*M*_r = 328.43
Orthorhombic, *Pbca*
a = 11.3660 (7) Å
b = 21.4096 (12) Å
c = 12.9034 (11) Å
V = 3139.9 (4) Å³
Z = 8
*D*_x = 1.390 Mg m⁻³

Mo Kα radiation
Cell parameters from 19865 reflections
 θ = 1.6–27.6°
 μ = 0.35 mm⁻¹
T = 293 K
Prism, colorless
0.38 × 0.35 × 0.29 mm

Data collection

Stoe IPDS-II diffractometer
 ω scans
Absorption correction: by integration (*X-RED32*; Stoe & Cie, 2002)
*T*_{min} = 0.877, *T*_{max} = 0.904
19955 measured reflections

3558 independent reflections
2155 reflections with *I* > 2σ(*I*)
*R*_{int} = 0.038
 θ _{max} = 27.5°
h = -14 → 14
k = -27 → 24
l = -16 → 15

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.055
wR (*F*²) = 0.178
S = 1.08
3558 reflections
214 parameters
H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.1017P)^2 + 0.0559P]$
where $P = (F_o^2 + 2F_c^2)/3$
(Δ/σ)_{max} < 0.001
 $\Delta\rho$ _{max} = 0.50 e Å⁻³
 $\Delta\rho$ _{min} = -0.37 e Å⁻³
Extinction correction: *SHELXL97*
Extinction coefficient: 0.0090 (18)

Table 1
Selected geometric parameters (Å, °).

S1—C3	1.716 (3)	N2—C5	1.388 (4)
S1—C4	1.725 (5)	N2—C6	1.397 (3)
S2—C9	1.779 (4)	N2—C3	1.351 (4)
S2—C10	1.815 (3)	N3—N4	1.389 (3)
O1—C7	1.218 (3)	N3—C7	1.360 (3)
O2—C8	1.218 (4)	N4—C8	1.337 (3)
N1—C2	1.384 (3)	N4—C10	1.469 (3)
N1—C3	1.319 (4)		
C3—S1—C4	89.08 (18)	N2—C5—C4	110.9 (4)
C9—S2—C10	93.04 (15)	N2—C6—C2	105.0 (2)
C2—N1—C3	105.0 (2)	N2—C6—C7	119.6 (2)
C3—N2—C6	106.8 (2)	N3—C7—C6	114.4 (2)
C5—N2—C6	139.1 (2)	O1—C7—C6	121.8 (2)
C3—N2—C5	114.1 (2)	O1—C7—N3	123.7 (2)
N4—N3—C7	117.9 (2)	N4—C8—C9	111.2 (3)
N3—N4—C10	117.8 (2)	O2—C8—N4	124.1 (2)
C8—N4—C10	119.6 (2)	O2—C8—C9	124.7 (3)
N3—N4—C8	120.4 (2)	S2—C9—C8	107.3 (2)
N1—C2—C6	110.5 (2)	S2—C10—C12	111.3 (2)
N1—C2—C1	119.4 (2)	S2—C10—C11	108.5 (3)
N1—C3—N2	112.8 (2)	N4—C10—C11	110.0 (2)
S1—C3—N2	111.8 (3)	S2—C10—N4	102.78 (17)
S1—C3—N1	135.4 (3)	N4—C10—C12	110.7 (3)
S1—C4—C5	114.1 (3)		

Table 2
Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N3—H3...O3 ⁱ	0.83 (4)	2.01 (4)	2.828 (3)	167 (3)
O3—H3A...N1 ⁱⁱ	0.91 (5)	1.89 (5)	2.795 (4)	174 (4)
O3—H3B...O2 ⁱⁱⁱ	0.78 (4)	2.16 (4)	2.871 (4)	152 (4)
C5—H5...O1	0.95 (4)	2.49 (3)	3.001 (4)	114 (3)
C9—H9B...O1 ^{iv}	0.97	2.29	3.225 (4)	163

Symmetry codes: (i) $\frac{1}{2} + x, y, \frac{3}{2} - z$; (ii) $x - 1, y, z$; (iii) $\frac{1}{2} - x, 1 - y, \frac{1}{2} + z$; (iv) $x - \frac{1}{2}, y, \frac{1}{2} - z$.

Methyl and methylene H atoms were positioned geometrically and constrained to an idealized geometry, with C—H distances of 0.96 Å for methyl and 0.97 Å for methylene groups. The *U*_{iso}(H) values were constrained to be 1.2 (1.5 for methyl group) times *U*_{eq} of the carrier atoms. The other H atoms were found in a difference Fourier map and refined isotropically.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s)

used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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